

## CYCLOHEXANEDIOL CRYOPROTECTANT COMPOUNDS

5        [0001] This invention was made with government support under grant number Cooperative Agreement Number 70NANB7H3071, awarded by the Department of Commerce. The government has certain rights in the invention.

### BACKGROUND OF THE INVENTION

#### 1.     Field of Invention

10        [0002] This invention relates to particular cyclohexanediol molecules and their use as cryoprotectants.

#### 2.     Description of Related Art

15        [0003] Cryobiology may be defined as the study of the effects of temperatures of lower than normal physiologic ranges upon biologic systems. During the past half-century the fundamentals of the science of cryobiology have evolved to the point where low temperatures are now used extensively as a means to protect and preserve biological systems during enforced periods of ischemia and hypoxia. In practice, preservation is achieved using either hypothermia without freezing, or cryopreservation in which the aqueous system sustains a physical phase change with the formation of ice. Survival of cells from the rigors of freezing and thawing in  
20        cryopreservation procedures is only attained by using appropriate cryoprotective agents (CPAs) and in general, these techniques are applicable to isolated cells in suspension or small aggregates of cells in simple tissues. More complex tissues and organs having a defined architecture are not easily preserved using conventional cryopreservation techniques, which is principally due to the deleterious effects of ice  
25        formation in an organized multicellular tissue. Simply freezing cells or tissues results in dead, nonfunctional materials.

30        [0004] The modern era of cryobiology really began with the discovery of the cryoprotective properties of glycerol as reported by Polge et al., "Revival of Spermatzoa After Vitrification and Dehydration at Low Temperatures," Nature, 164:666 (1949). Subsequently, Lovelock et al., "Prevention of Freezing Damage to Living Cells by Dimethyl Sulfoxide," Nature, 183:1394 (1959), discovered that dimethyl sulfoxide was also a cryoprotectant, and despite the wide range of

compounds now known to exhibit cryoprotective properties, it is still the most widely used compound to date.

[0005] A review of the principles of cryobiology can be found in Brockbank, Principles of Cryopreserved Venous Transplantation, Chapter 10, "Essentials of Cryobiology" (1995). A basic principle of cryobiology is that the extent of freezing damage depends upon the amount of free water in the system and the ability of that water to crystallize during freezing. Many types of isolated cells and small aggregates of cells can be frozen simply by following published procedures, but obtaining reproducible results for more complex tissues requires an understanding of the major variables involved in tissue cryopreservation. Major variables involved in tissue freezing include (1) freezing-compatible pH buffers, (2) cryoprotectant choice, concentration and administration, (3) cooling protocol, (4) storage temperature, (5) warming protocol and (6) cryoprotectant elution.

[0006] Many cryoprotectants have been discovered. See, for example, Brockbank, *supra*. Cryoprotectant selection for cryopreservation is usually restricted to those that confer cryoprotection in a variety of biological systems. On occasion, combinations of cryoprotectants may result in additive or synergistic enhancement of cell survival. Comparison of chemicals with cryoprotectant properties reveals no common structural features. These chemicals are usually divided into two classes: (1) intracellular cryoprotectants with low molecular weights that penetrate cells, and (2) extracellular cryoprotectants with relatively high molecular weights (greater than or equal to sucrose (342 daltons)) which do not penetrate cells. Intracellular cryoprotectants, such as glycerol and dimethyl sulfoxide at concentrations from 0.5 to 3 molar, are effective in minimizing cell damage in many slowly frozen biological systems. Extracellular cryoprotective agents such as polyvinylpyrrolidone or hydroxyethyl starch are often more effective at protecting biological systems cooled at rapid rates.

[0007] What is still desired are improved cryoprotectant materials that increase cell viability during cryopreservation.

### SUMMARY OF THE INVENTION

[0008] It is therefore one object of the present invention to provide a cryoprotectant material that effectively protects cells during cryopreservation and achieves increased cell viability upon warming from a frozen state.

[0009] It is still a further object of the present invention to provide a cryoprotectant material that is capable of obtaining consistent and reproducible results in cryopreserving cells and tissues.

[0010] It is a still further object of the present invention to provide a cryoprotectant material that is able to work in conjunction with naturally occurring anti-freeze proteins (AFPs) to promote survival of cells after freezing in a cumulative manner.

[0011] These and other objects are achieved by the present invention, which relates to the use of newly discovered cryoprotectant compounds. In particular, the invention relates to the use of cyclohexanediol compounds, specifically the cis or trans forms of 1,3-cyclohexanediol (1,3CHD) and 1,4-cyclohexanediol (1,4CHD), and their racemic mixtures, as cryoprotectants in preserving living cells.

[0012] In the invention, cells to be cryopreserved are protected against the effects of cryopreservation by bringing the cells into contact with a cryopreservation composition containing at least one cyclohexanediol compound, and subsequently reducing the temperature of the cells to the cryopreservation temperature.

[0013] Also in the invention, the cryopreservation composition preferably comprises not only at least one cyclohexanediol compound, but also at least one additional cryoprotectant compound and/or at least one anti-freeze protein.

### BRIEF DESCRIPTION OF THE DRAWINGS

[0014] Figure 1 is a flow chart summarizing the cryopreservation procedure utilized in obtaining the results summarized in this application.

[0015] Figures 2-3 are plots of relative cell viability after freezing using CHD compounds in conjunction with conventional cryoprotective agents.

[0016] Figures 4-5 are plots of relative cell viability after freezing using CHD compounds in conjunction with anti-freeze proteins.

### DETAILED DESCRIPTION OF PREFERRED EMBODIMENTS

5 [0017] The inventors have discovered two new compounds exhibiting cryoprotective activity, 1,3-cyclohexanediol (1,3CHD) and 1,4-cyclohexanediol (1,3CHD). The inventors have also discovered that these compounds are able to work in conjunction with naturally occurring anti-freeze proteins (AFPs) to promote survival after freezing in a cumulative manner.

[0018] Cryopreservation, i.e., the preservation of cells by freezing, in the present invention may be effected in any conventional manner. By "freezing" as used herein is meant temperatures below the freezing point of water, i.e., below 0°C.

10 Cryopreservation typically involves freezing cells to temperatures well below freezing, for example to -130°C or less. The cryopreservation temperature should be less than -20°C, more preferably -80°C or less, most preferably -130°C or less.

[0019] The cells to be cryopreserved using the CHD cryoprotectant compounds of the invention may be in suspension, may be attached to a substrate,

15 etc., without limitation.

[0020] In the method of the invention, the cells to be protected during cryopreservation are first brought into contact with a cryopreservation composition. By being brought into contact with the cryopreservation composition is meant that the cells are made to be in contact in some manner with the cryopreservation composition

20 so that during the reduction of temperature to the cryopreservation temperature, the cells are protected by the cryopreservation composition. For example, the cells may be brought into contact with the cryopreservation composition by filling the appropriate wells of a plate to which the cells to be protected are attached, by suspending the cells in a solution of the cryopreservation composition, etc.

25 [0021] The cells to be cryopreserved should also preferably be in contact with freezing compatible pH buffer comprised most typically of at least a basic salt solution, an energy source (for example, glucose) and a buffer capable of maintaining a neutral pH at cooled temperatures. Well known such materials include, for example, Dulbecco's Modified Eagle Medium (DMEM). This material may also be included as

30 part of the cryopreservation composition.

[0022] The cryopreservation composition of the invention must contain at least one cyclohexanediol (CHD) compound, for example the cis or trans forms of

1,3-cyclohexanediol or 1,4-cyclohexanediol and racemic mixtures thereof.

Preferably, the CHD compound is present in the cryopreservation composition in an amount of from, for example, 0.05 to 2.0 M, more preferably from 0.1 M to 1.0 M.

5 [0023] The cryopreservation composition also preferably includes a solution well suited for organ storage. The solution can include the buffers discussed above. A particularly preferred solution is, for example, EuroCollins Solution comprised of dextrose, potassium phosphate monobasic and dibasic, sodium bicarbonate and potassium chloride.

10 [0024] In a further embodiment of the invention, the cryopreservation composition contains not only the CHD compound, but also at least one additional cryoprotectant compound. These additional cryoprotectant compounds may include, for example, any of those set forth in Table 10.1 of Brockbank, supra, including, but not limited to, acetamide, agarose, alginate, l-analine, albumin, ammonium acetate, butanediol, chondroitin sulfate, chloroform, choline, dextrans, diethylene glycol, 15 dimethyl acetamide, dimethyl formamide, dimethyl sulfoxide (DMSO), erythritol, ethanol, ethylene glycol, formamide, glucose, glycerol,  $\alpha$ -glycerophosphate, glycerol monoacetate, glycine, hydroxyethyl starch, inositol, lactose, magnesium chloride, magnesium sulfate, maltose, mannitol, mannose, methanol, methyl acetamide, methylformamide, methyl ureas, phenol, pluronic polyols, polyethylene glycol, 20 polyvinylpyrrolidone, proline, propylene glycol, pyridine N-oxide, ribose, serine, sodium bromide, sodium chloride, sodium iodide, sodium nitrate, sodium sulfate, sorbitol, sucrose, trehalose, triethylene glycol, trimethylamine acetate, urea, valine, xylose, etc. This additional cryoprotectant compound is preferably present in the cryopreservation composition in an amount of from, for example, 0.1 M to 10.0 M, 25 preferably 0.1 to 2.0 M.

[0025] In a still further embodiment of the invention, the cryopreservation composition includes the CHD compound, with or without an additional cryoprotectant compound, and also includes an anti-freeze protein/peptide (AFP). AFPs also include anti-freeze glycoproteins (AFGPs) and insect anti-freeze, or "thermal 30 hysteresis" proteins, (THPs). Naturally occurring AFPs are believed to be able to bind to the prism face of developing ice crystals, thereby altering their formation. For the fishes and insects in which these proteins occur, it means a depression of their

freezing point so they are able to survive under conditions that would normally cause their body fluids to freeze.

[0026] Any of the well-known AFPs may be used in the present invention in this regard. See, for example, Sicheri and Yang, Nature, 375:427-431, (1995),  
 5 describing eight such proteins. Most preferably, the AFP may be, for example, AFPI (AFP type I), AFPIII (AFP type III) and/or AFGP.

[0027] The AFPs may be present in the cryopreservation composition in an amount of from, for example, 0.01 to 1 mg/mL, more preferably 0.05 to 0.5 mg/mL, of composition, for each AFP present.

10 [0028] Once the cells have been contacted with the cryopreservation composition, the cells may then be frozen for cryopreservation. The cryopreservation and subsequent warming of cells may be conducted in any manner, and may utilize any additional materials, well known in the art. Preferred embodiments are described in the following discussion and the Examples set forth below.

15 [0029] The cooling (freezing) protocol for cryopreservation in the present invention may be any suitable type. Many types of cooling protocols are well known to practitioners in the art. Most typically, the cooling protocol calls for continuous rate cooling from the point of ice nucleation to  $-80^{\circ}\text{C}$ , with the rate of cooling depending on the characteristics of the cells/tissues being frozen as understood in the  
 20 art (again, see Brockbank, *supra*). The cooling rate may be, for example,  $-0.1^{\circ}\text{C}$  to  $-10^{\circ}\text{C}$  per minute, more preferably between  $-1^{\circ}\text{C}$  to  $-2^{\circ}\text{C}$  per minute. Once the cells are cooled to about  $-80^{\circ}\text{C}$  by this continuous rate cooling, they can be transferred to liquid nitrogen or the vapor phase of liquid nitrogen for further cooling to the  
 25 cryopreservation temperature, which is below the glass transition temperature of the freezing solution (again, typically  $-130^{\circ}\text{C}$  or less).

[0030] Once cryopreserved, the cells will subsequently be rewarmed for removal of the cryopreserved cells from the cryopreserved state. The warming protocol for taking the cells out of the frozen state may be any type of warming protocol, which are well known to practitioners in the art. Typically, the warming is  
 30 done in a one-step procedure in which the cryopreserved specimen is placed into a water bath (temperature of about  $37-42^{\circ}\text{C}$ ) until complete rewarming is effected. More rapid warming is also known.

[0031] Most preferably, the cryopreserved cells, particularly cryopreserved cells fixed to a substrate, are warmed by way of the methods described in co-pending Application No. \_\_\_\_\_ (Docket No. 106006) filed on even date herewith, entitled "Novel Warming Method of Cryopreserved Specimens," incorporated herein by reference in its entirety. These methods include a two-step warming protocol, with or without the use of a heat sink.

[0032] The cryopreservation composition of the present invention that includes at least one CHD compound is surprisingly able to increase the survival of cryopreserved cells upon freezing in a cumulative manner. The following examples illustrate the surprising utility of the CHD compounds as a cryoprotectant.

### EXAMPLES

#### Example 1

[0033] A primary cell strain called AV5 was used for these experiments. AV5 cells are derived from porcine heart valve leaflets. Hearts were obtained from pigs and the heart valve leaflets were then removed and washed several times with sterile phosphate-buffered saline (PBS). Small pieces (~1 mm<sup>2</sup>) were cut and placed into a 24-well microtiter plate coated with 0.2% gelatin. Dulbecco's Modified Eagle's Medium (DMEM) with 10% fetal calf serum (FCS) was added to just cover the bottom of the well and the plate was left at 37°C with 5% CO<sub>2</sub> in air until visible outgrowth occurred.

[0034] Outgrowth was allowed to continue until cells filled the well at which time the cells were removed from the well using trypsin and placed into a small tissue culture flask. Once the flask reached confluency, the cells were again removed with trypsin and stored in 10% dimethyl sulfoxide (DMSO) at -135°C.

[0035] To evaluate the cryoprotective capabilities of the CHD compounds, the protocol of Figure 1 was followed. AV5 cells were plated the day before each experiment at 25,000 cells/well. At the beginning of each experiment, the plate was placed on ice and the cells were exposed to mannitol prior to loading the various cryopreservation compositions.

[0036] All of the cryopreservation compositions were formulated in EuroCollins Solution, consisting of dextrose, potassium phosphate monobasic and dibasic, sodium bicarbonate and potassium chloride. The plates were then cooled at

the rate of  $-1.0^{\circ}\text{C}/\text{min}$  to  $-80^{\circ}\text{C}$ , and then further cooled with liquid nitrogen vapor and stored overnight at  $-135^{\circ}\text{C}$ .

[0037] The next day the plate was removed and warmed to  $\sim 4^{\circ}\text{C}$  at which point it was put back on ice. During warming,  $150\ \mu\text{l}$  of  $0.5\ \text{M}$  mannitol in cell culture media was added to the cells. Once back on ice, the cryoprotectant/mannitol mixture was removed. The cells were washed twice with  $0.5\ \text{M}$  mannitol/media and twice with DMEM (10%FCS).

[0038] Cell viability was then determined using the non-invasive metabolic indicator Alamar Blue (Trek Diagnostics). Alamar Blue is a fluorescent dye that measures the oxidation/reduction reactions within cells, and thus is indicative of the overall viability of the cells after exposure to cryoprotective agents. A volume of  $20\ \mu\text{l}$  Alamar Blue was added to cells in  $200\ \mu\text{l}$  of DMEM (10%FCS) and the plate was incubated at  $37^{\circ}\text{C}$  for 3 hours. Fluorescence from Alamar Blue was read in a fluorescent microplate reader (Fmax fluorescent microplate reader by Molecular Dynamics) using an excitation wavelength of  $544\ \text{nm}$  and an emission wavelength of  $590\ \text{nm}$ .

[0039] The first set of experiments involved using two CPA compositions, either  $1\ \text{M}$  dimethyl sulfoxide (DMSO) (left bar of graph at each concentration of 1,3CHD in Figure 2 and 1,4CHD in Figure 3) or a combination of DMSO, formamide and propanediol at a final concentration of  $1\ \text{M}$  (right bar of graph at each concentration of 1,3CHD in Figure 2 and 1,4CHD in Figure 3). Concentrations varying from  $0$  to  $1\ \text{M}$  of 1,3CHD and  $0$  to  $1\ \text{M}$  1,4CHD were added to these two separate CPA compositions for additional experiments. Cell viability was assessed using the assay described above.

[0040] The results are graphically summarized in Figures 2 and 3. Figure 2 relates to 1,3CHD, while Figure 3 relates to 1,4CHD. For each, the data was normalized to the conventional cryoprotectant alone and is the mean ( $\pm$  SEM) of 12 replicates. As shown in these two Figures, in the presence of varying concentrations of both CHD molecules, viability was significantly increased over the comparative cryopreservation compositions that contained only the conventional cryoprotectants without a CHD compound (i.e., concentration of  $0.00\ \text{CHD}$ ).



[0041] Similar results to the foregoing for AV5 have been obtained with other cell types, including (1) A10, an established cell line of smooth muscle cells derived from rat thoracic aorta and (2) J15, a primary cell strain of smooth muscle cells derived from rabbit jugular veins.

## 5            Example 2

[0042] In this Example, an anti-freeze protein (AFP) was added to the cryoprotective composition. Varying concentrations of three different AFPs (AFPI, AFPIII and AFGP) were used along with 1 M DMSO and either 0.25 M 1,3CHD (Figure 5) or 0.5 M 1,4CHD (Figure 4). Again, the protocol of Figure 1 was followed. The data were normalized to the comparative conventional cryoprotectant alone (i.e., DMSO alone), and the results are presented in Figures 4 and 5.

[0043] In Figure 4, the relative cell viability of AV5 cells after freezing using a combination of 1,4CHD/AFPI/DMSO is summarized. Concentrations of the constituents included 0.5 M 1,4CHD, 0.1 mg/mL AFPI and 1 M DMSO. The graph depicts viability in the presence of 1 M DMSO alone or in combination with 1,4CHD or 1,4CHD plus AFPI at the above concentrations. Data was normalized to the conventional cryoprotectant (DMSO) alone and is the mean (+/- SEM) of 3 replicates.

[0044] In Figure 4, the results demonstrated the increased viability of the cells upon addition of the CHD molecule as compared to the conventional cryoprotectant alone and a further increase in viability upon addition of the AFPI protein to the conventional cryoprotectant/CHD mixture. Thus, a cumulative effect in the presence of AFP and CHD was demonstrated.

[0045] In Figure 5, the relative cell viability of AV5 cells after freezing using a combination of 1,3CHD, DMSO and three different AFP proteins is summarized. Concentrations of the constituents included 0.25 M 1,3CHD, 1 M DMSO and 0.1 mg/mL for each AFP protein (left bar of graph is with AFPI, middle bar of graph is with AFPIII, and right bar of graph is with AFGP). Data was normalized to the DMSO alone and is the mean (+/- SEM) of 3 replicates.

[0046] From the data of Figure 5, AFPI appears to confer the best protection to the cells in this Example as observed by an increase in cell viability as compared to the conventional cryoprotectant alone or to the conventional cryoprotectant/CHD

mixture. This cumulative increase in viability is not observed in the presence of the AFPs plus a conventional cryoprotectant alone (i.e., without the CHD present).

5 [0047] While this invention has been described in conjunction with specific embodiments thereof, it is evident that many alternatives, modifications and variations will be apparent to those skilled in the art. Accordingly, the preferred embodiments of the invention as set forth herein are intended to be illustrative only, and not limiting. Various changes may be made without departing from the spirit and scope of the invention as defined in the following claims.